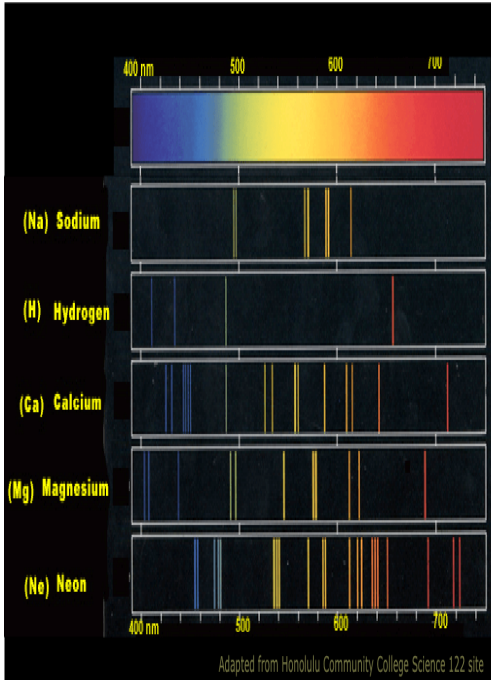




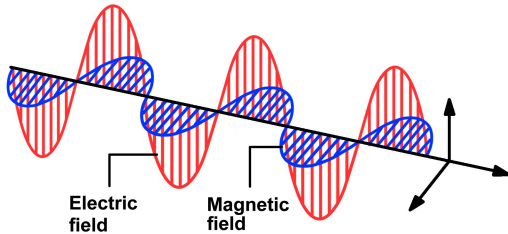
Atomic spectroscopy



Adapted from Honolulu Community College Science 122 site

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I. Electromagnetic radiation	2
I.A. Electromagnetic waves	2
I.B. Photons	3
I.C. EM spectrum	4
II. Matter-radiation interaction	6
II.A. Time-dependent Schrödinger's equation	6
II.B. Transition probability	7
III. Atomic spectroscopy	13
III.A. Hydrogen atom	13



■ Electric and magnetic fields

$$\mathbf{E} = \mathbf{i}E_x = \mathbf{i}E_{0x} \cos(\omega t - kz)$$

$$\mathbf{B} = \mathbf{j}B_y = \mathbf{j}B_{0y} \cos(\omega t - kz)$$

■ Periodicity

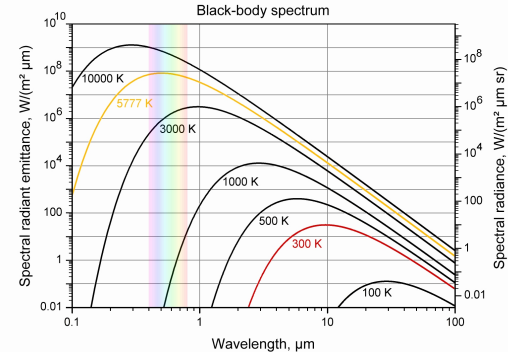
- Space $\Rightarrow E(z) = E(z + \lambda) \Rightarrow k = 2\pi/\lambda$
- Time $\Rightarrow E(t) = E(t + \tau) \Rightarrow \omega = 2\pi/\tau = 2\pi\nu$

■ Velocity $\Rightarrow c = \frac{1}{\sqrt{\epsilon_0\mu_0}} = 2.9979 \cdot 10^8 \text{ m/s}$



I.B. Photons

- Plank (1900) \Rightarrow EM radiation \Rightarrow Photons \Rightarrow Einstein (1905)

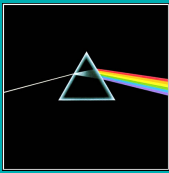


- Energy $\Rightarrow E = h\nu \Rightarrow E = h\frac{c}{\lambda} = \hbar\omega$ ($\hbar = \frac{h}{2\pi}$)
- Momentum

$$E = \sqrt{m_0^2 c^4 + p^2 c^2}$$

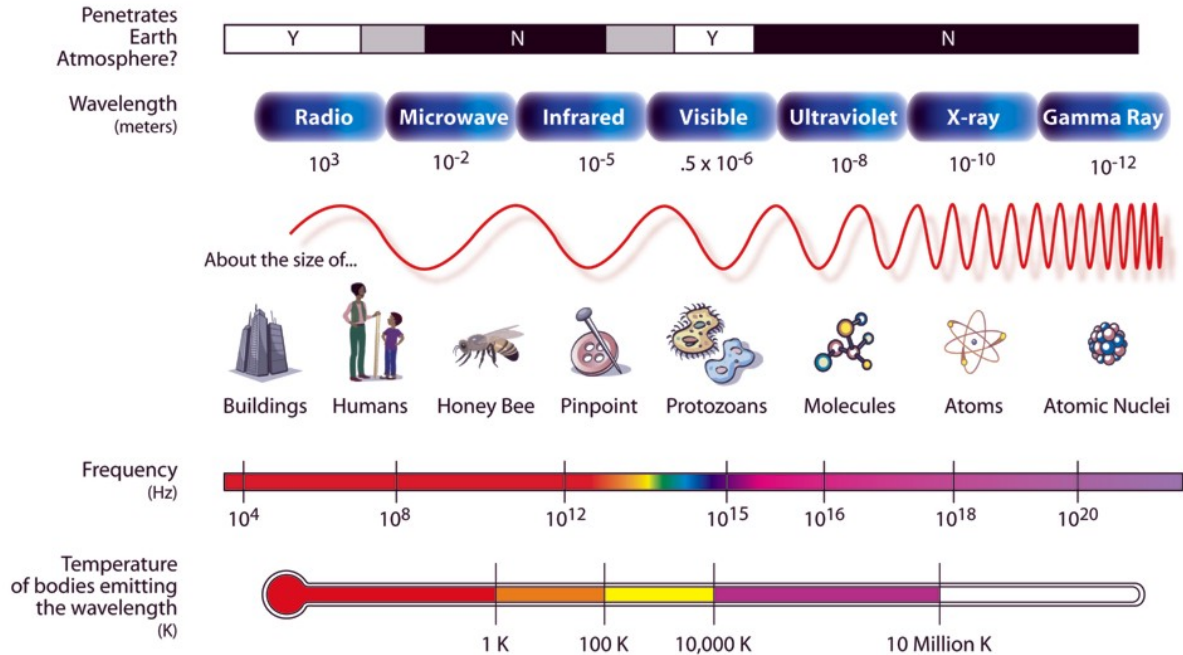
$$\Downarrow m_0=0$$

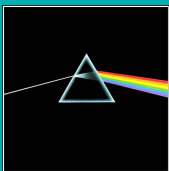
$$E = pc \rightarrow p = \frac{E}{c} = \frac{h}{\lambda}$$



I.C. EM spectrum

THE ELECTROMAGNETIC SPECTRUM





I.C. EM spectrum

EM radiation	λ (nm)	E (eV)	Molecular transitions
Radio	$3 \times 10^8 -$	$4 \times 10^{-6} -$	
Microwaves	$3 \times 10^8 - 10^6$	$4 \times 10^{-6} - 1.2 \times 10^{-3}$	rotation
Infrared (IR)	$3.84 \times 10^{14} - 10^6$	$1.2 \times 10^{-3} - 1.7$	vibration
Visible	780 - 390	1.7 - 3.2	electronic
Ultraviolet (UV)	390 - 10	3.2 - 120	electronic
X Rays	10 - 0.006	120 - 2.4×10^5	ionization
Gamma Rays	0.1 - 1×10^{-5}	$10^4 - 10^8$	



II.A. Time-dependent Schrödinger's equation

- Postulate 5. The time evolution of the system is governed by the time-dependent Schrödinger's equation

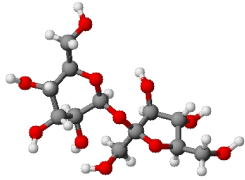
$$i\hbar \frac{\partial \Psi(x,t)}{\partial t} = \hat{H}(x,t) \Psi(x,t)$$

- Stationary state

$$\Psi(x,t) = \psi(x) e^{-iEt/\hbar} \rightarrow \begin{cases} |\Psi(x,t)|^2 = |\psi(x)|^2 \\ \hat{H}\psi = E\psi \end{cases}$$



II.B. Transition probability



- Isolated system $\Rightarrow \hat{H}^{(0)}(x)\psi_n^{(0)}(x) = E_n^{(0)}\psi_n^{(0)}(x)$

- Stationary state

$$\psi(x,t) = \Psi_n^{(0)}(x)e^{-iE_n^{(0)}t/\hbar} \rightarrow |\Psi(x,t)|^2 = |\Psi_n^{(0)}(x)|^2$$

- Perturbation (EM) $\Rightarrow \hat{H}(x,t) = \hat{H}^{(0)}(x) + \hat{H}'(x,t)$

$$i\hbar \frac{\partial \Psi(x,t)}{\partial t} = \hat{H}(x,t)\Psi(x,t) \rightarrow \Psi(x,t) = \sum_j c_j(t)\psi_j^{(0)}(x)e^{-iE_j^{(0)}t/\hbar}$$

$$i\hbar \sum_j \frac{dc_j(t)}{dt} \psi_j^{(0)}(x)e^{-iE_j^{(0)}t/\hbar} = \sum_j c_j(t)\hat{H}'(x,t)\psi_j^{(0)}(x)e^{-iE_j^{(0)}t/\hbar}$$

$$\int \forall x (\psi_j^{(0)}(x))^* \psi_m^{(0)}(x) dx = \langle \psi_j^{(0)} | \psi_m^{(0)} \rangle = \delta_{j,m} \begin{cases} = 1 & m = j \\ = 0 & m \neq j \end{cases}$$



II.B. Transition probability

$$i\hbar \frac{dc_m(t)}{dt} = \sum_j c_j(t) H'_{mj}(t) e^{i\omega_{mj}t} \quad m=1,2,3,\dots \left\{ \begin{array}{l} H'_{mj}(t) = \langle \Psi_m^{(0)} | \hat{H}'(x,t) | \Psi_j^{(0)} \rangle \\ \omega_{mj} = \frac{E_m^{(0)} - E_j^{(0)}}{\hbar} \end{array} \right.$$

⇓

$$\Psi(x, 0) = \Psi_n^{(0)}(x) \rightarrow c_j(0) = \delta_{n,j}$$

- If $H' \ll E_m^{(0)} - E_n^{(0)} \Rightarrow i\hbar \frac{dc_m(t)}{dt} \simeq H'_{mn}(t) e^{i\omega_{mn}t}$

$$c_m(t) = \frac{1}{i\hbar} \int_0^t H'_{mn}(t') e^{i\omega_{mn}t'} dt'$$

$$P_{n \rightarrow m}(t) = |c_m(t)|^2 = \frac{1}{\hbar^2} \left| \int_0^t H'_{mn}(t') e^{i\omega_{mn}t'} dt' \right|^2$$



II.B. Transition probability

- Force

$$\mathbf{F} = q\mathbf{E} + \frac{q}{c}\mathbf{v} \times \mathbf{B}$$

$$\Downarrow \langle v^2 \rangle_{1s}^{1/2} \sim \frac{c}{137}$$

$$\mathbf{F} \simeq q\mathbf{E}$$

- Potential energy

$$\mathbf{F} = -\nabla V \rightarrow V = -\int \mathbf{F} \, d\mathbf{r}$$

\Downarrow discrete charges

$$V = -\underbrace{\sum_i q_i \mathbf{r}_i \cdot \mathbf{E}}_{\boldsymbol{\mu}} = -\boldsymbol{\mu} \cdot \mathbf{E}$$

$\boldsymbol{\mu} \rightarrow$ electric dipole moment



II.B. Transition probability

- Dipolar approximation \Rightarrow molecular size $\ll \lambda$

$$\mathbf{E} \simeq \mathbf{i}E_{0x} \cos(\omega t) \Rightarrow H' = -\boldsymbol{\mu} \cdot \mathbf{E} = -\mu_x E_{0x} \cos(\omega t)$$

$$H'_{mn}(t) = -E_{0x} \underbrace{\langle m | \mu_x | n \rangle}_{\substack{\text{transition} \\ \text{dipole} \\ \text{moment}}} \cos(\omega t) \leftarrow \mu_x = \sum_i q_i x_i$$

↓

$$P_{n \rightarrow m}(t) = \frac{|E_{0x}|^2 |\langle m | \mu_x | n \rangle|^2}{\hbar^2} \left| \int_0^t \cos \omega t' e^{i\omega_{mn} t'} dt' \right|^2$$



II.B. Transition probability

$$P_{n \rightarrow m}(t) = \frac{|E_{0x}|^2 |\langle m | \mu_x | n \rangle|^2}{4\hbar^2} \left| \frac{e^{i(\omega_{mn} + \omega)t} - 1}{\omega_{mn} + \omega} + \frac{e^{i(\omega_{mn} - \omega)t} - 1}{\omega_{mn} - \omega} \right|^2$$

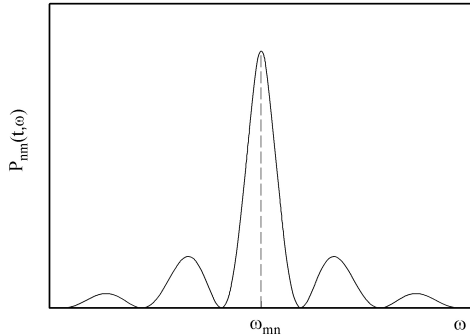
- $E_{0x} \neq 0$
- $\langle m | \mu_x | n \rangle \neq 0 \Rightarrow$ Selection rules
- If $\omega_{mn} - \omega \ll \omega_{mn} + \omega$

$$P_{n \rightarrow m}(t) = \frac{|E_{0x}|^2 |\langle m | \mu_x | n \rangle|^2}{4\hbar^2} \left| \frac{e^{i(\omega_{mn} - \omega)t} - 1}{\omega_{mn} - \omega} \right|^2$$

$$P_{n \rightarrow m}(t) = \frac{|E_{0x}|^2 |\langle m | \mu_x | n \rangle|^2 \text{sen}^2(\omega_{mn} - \omega)t/2}{\hbar^2 (\omega_{mn} - \omega)^2}$$



II.B. Transition probability



- Maximun

$$\omega = \omega_{mn} \rightarrow E_m - E_n = h\nu$$

- Absortion

$$\nu > 0 \Rightarrow E_m > E_n$$

- If $\omega_{mn} + \omega \ll \omega_{mn} - \omega$

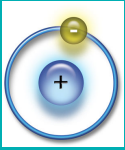
$$P_{n \rightarrow m}(t) = \frac{|E_{0x}|^2 |\langle m | \mu_x | n \rangle|^2 \text{sen}^2(\omega_{mn} + \omega)t/2}{\hbar^2 (\omega_{mn} + \omega)^2}$$

- Maximun

$$\omega = -\omega_{mn} \rightarrow E_n - E_m = h\nu$$

- Emission

$$\nu > 0 \Rightarrow E_n > E_m$$



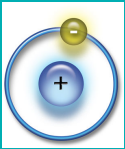
III.A. Hydrogen atom

- Transition dipole moment $\Rightarrow \langle \Psi_i | \boldsymbol{\mu} | \Psi_j \rangle$

$$\left\{ \begin{array}{l} \boldsymbol{\mu} = -e \mathbf{r} \\ \mathbf{r} = x\mathbf{i} + y\mathbf{j} + z\mathbf{k} = r \sin \theta \cos \phi \mathbf{i} + r \sin \theta \sin \phi \mathbf{j} + r \cos \theta \mathbf{k} \\ \langle \Psi_i | \boldsymbol{\mu} | \Psi_j \rangle = \langle \Psi_{n',l',m',m'_s} | \boldsymbol{\mu} | \Psi_{n,l,m,m_s} \rangle \end{array} \right.$$

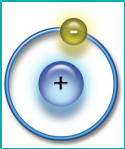
$$\begin{aligned} \langle \Psi_{n',l',m',m'_s} | \boldsymbol{\mu} | \Psi_{n,l,m,m_s} \rangle &= -e \int_0^\infty R_{n',l'}^*(r) r^3 R_{n,l}(r) dr \\ &\cdot \int_0^{2\pi} \int_0^\pi \Theta_{l',m'}^*(\theta) \Phi_{m'}^*(\phi) [\sin \theta \cos \phi \mathbf{i} + \sin \theta \sin \phi \mathbf{j} \\ &+ \cos \theta \mathbf{k}] \Theta_{l,m}(\theta) \Phi_m(\phi) \sin \theta d\theta d\phi \langle m'_s | m_s \rangle \neq 0 \end{aligned}$$

- $\langle m'_s | m_s \rangle \neq 0 \Rightarrow m_s = m'_s \Rightarrow \Delta m_s = 0$



III.A. Hydrogen atom

- $\langle \Phi_{m'}^* | \Phi_m \rangle = \int_0^{2\pi} e^{-im'\phi} e^{im\phi} d\phi \neq 0 \Rightarrow m=m' \Rightarrow \Delta m=0$
- $\langle \Phi_{m'}^* | \sin \phi | \Phi_m \rangle = \int_0^{2\pi} e^{-im'\phi} \sin \phi e^{im\phi} d\phi \neq 0 \Rightarrow m'=m+1 \Rightarrow \Delta m=+1$
- $\langle \Phi_{m'}^* | \cos \phi | \Phi_m \rangle = \int_0^{2\pi} e^{-im'\phi} \cos \phi e^{im\phi} d\phi \neq 0 \Rightarrow m'=m-1 \Rightarrow \Delta m=-1$
- $\left\{ \begin{array}{l} \langle \Theta_{l'm'}^* | \cos \theta | \Theta_{lm} \rangle = \int_0^{2\pi} \Theta_{l'm'}^*(\theta) \cos \theta \Theta_{lm}(\theta) \sin \theta d\theta \neq 0 \\ \langle \Theta_{l'm'}^* | \sin \theta | \Theta_{lm} \rangle = \int_0^{2\pi} \Theta_{l'm'}^*(\theta) \sin \theta \Theta_{lm}(\theta) \sin \theta d\theta \neq 0 \end{array} \right\} \Rightarrow \Delta l = \pm 1$
- $\langle R_{n',l'}^* | r | R_{n,l} \rangle = \int_0^\infty R_{n',l'}^*(r) r^3 R_{n,l}(r) dr \neq 0 \Rightarrow \Delta n = 0, \pm 1, \pm 2, \dots$

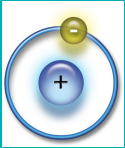


III.A. Hydrogen atom

- Energy differences (excitation $\Rightarrow n_2 > n_1$)

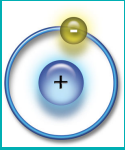
$$\Delta E_{n_1 \rightarrow n_2} = E_{n_2} - E_{n_1} = -\frac{\mu Z^2 k^2 e^4}{2\hbar^2} \left(\frac{1}{n_2^2} - \frac{1}{n_1^2} \right)$$

- Lyman series $\Rightarrow n_1 = 1 \Rightarrow$ transitions from the $1s$ state
 - $1s \rightarrow 2s \Rightarrow \Delta l = 0 \Rightarrow$ Forbidden
 - $1s \rightarrow 2p \Rightarrow \Delta l = 1 \Rightarrow$ Allowed
 - $1s \rightarrow 3s \Rightarrow \Delta l = 0 \Rightarrow$ Forbidden
 - $1s \rightarrow 3p \Rightarrow \Delta l = 1 \Rightarrow$ Allowed
 - $1s \rightarrow 3d \Rightarrow \Delta l = 2 \Rightarrow$ Forbidden
 - \vdots
 - Transitions $1s \rightarrow np$



III.A. Hydrogen atom

- Balmer series $\Rightarrow n_1 = 2 \Rightarrow$ transitions from the $2s$ or $2p$ states
 - $2s \rightarrow 3s \Rightarrow \Delta l = 0 \Rightarrow$ Forbidden
 - $2s \rightarrow 3p \Rightarrow \Delta l = 1 \Rightarrow$ Allowed
 - $2s \rightarrow 3d \Rightarrow \Delta l = 2 \Rightarrow$ Forbidden
 - $2p \rightarrow 3s \Rightarrow \Delta l = -1 \Rightarrow$ Allowed
 - $2p \rightarrow 3p \Rightarrow \Delta l = 0 \Rightarrow$ Forbidden
 - $2p \rightarrow 3d \Rightarrow \Delta l = 1 \Rightarrow$ Allowed
 - \vdots
 - Degenerated transitions $2s \rightarrow np, 2p \rightarrow ns, 2p \rightarrow nd$



III.A. Hydrogen atom

- Initial state $2s \Rightarrow$ Relaxation into $1s \Rightarrow \Delta l = 0 \Rightarrow$ Forbidden \Rightarrow Metastable state
- Spin-orbit coupling $\Rightarrow \propto \mathbf{L} \cdot \mathbf{S}$