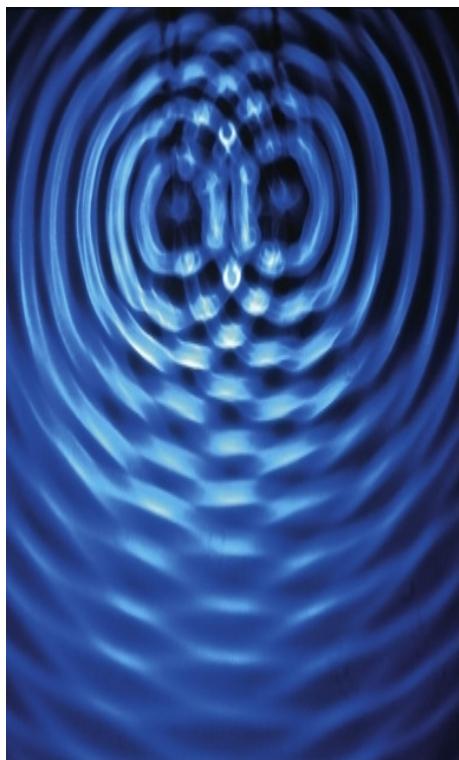


Useful quantum models in Chemistry

TEMA: USEFUL QUANTUM MODELS
IN CHEMISTRY

Introduction



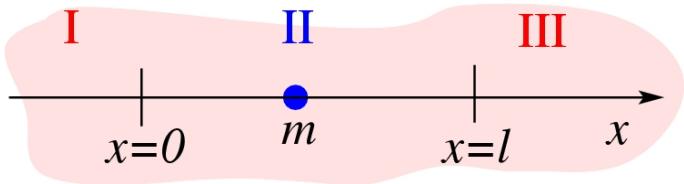
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I.A. Schrödinger's equation

USEFUL QUANTUM MODELS IN CHEMISTRY

I. Particle in a box

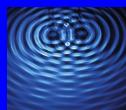


$$V(x) = \begin{cases} +\infty & x \leq 0 \\ 0 & 0 < x < l \\ +\infty & l \leq x \end{cases}$$

- Zones (I) and (III) $\Rightarrow |\psi(x)|^2 = 0 \rightarrow \psi_I(x) = \psi_{III}(x) = 0$
 - Zone (II)

$$\hat{H}\Psi = E\Psi$$

$$-\frac{\hbar^2}{2m} \frac{d^2\psi(x)}{dx^2} + \underbrace{V(x)\psi(x)}_0 = E\psi(x)$$



I.A. Schrödinger's equation

USEFUL QUANTUM MODELS IN CHEMISTRY

I. Particle in a box

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$$\frac{d^2\psi(x)}{dx^2} = -\frac{2mE}{\hbar^2}\psi(x)$$

- Independent solutions

$$\psi_1(x) = \sin(kx)$$

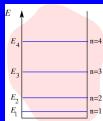
$$\psi_2(x) = \cos(kx)$$

$$\frac{d^2\psi_1(x)}{dx^2} = -k^2 \sin(kx)$$

$$\frac{d^2\psi_2(x)}{dx^2} = -k^2 \cos(kx)$$

$$k = \frac{\sqrt{2mE}}{\hbar}$$

$$\psi(x) = A\psi_1(x) + B\psi_2(x) = A \sin\left(\frac{\sqrt{2mE}}{\hbar}x\right) + B \cos\left(\frac{\sqrt{2mE}}{\hbar}x\right)$$



I.B. Eigenfunctions and eigenvalues

$$\psi(x) = A \sin\left(\frac{\sqrt{2mE}}{\hbar}x\right) + B \cos\left(\frac{\sqrt{2mE}}{\hbar}x\right)$$

- Continuity $x = 0 \Rightarrow \psi_{\text{I}}(x = 0) = \psi_{\text{II}}(x = 0)$

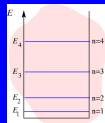
$$A \sin\left(\frac{\sqrt{2mE}}{\hbar}0\right) + B \cos\left(\frac{\sqrt{2mE}}{\hbar}0\right) = 0$$

$$A \cdot 0 + B \cdot 1 = 0$$

$$B = 0$$

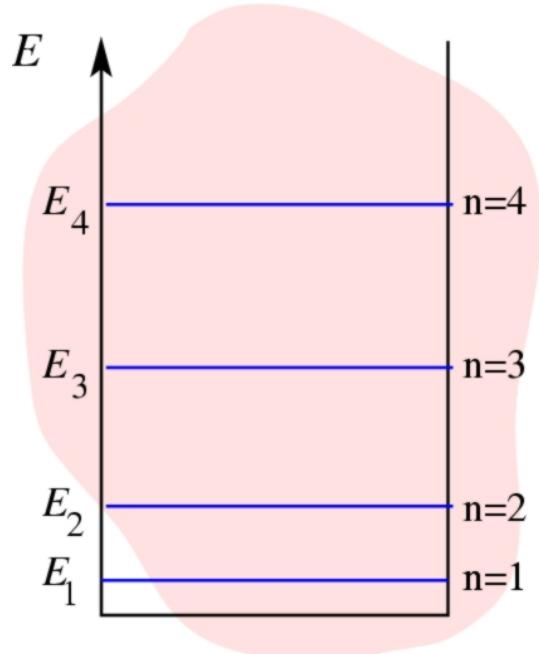
- Continuity $x = l \Rightarrow \psi_{\text{II}}(x = l) = \psi_{\text{III}}(x = l)$

$$A \cdot \sin\left(\frac{\sqrt{2mE}}{\hbar}l\right) = 0 \Rightarrow \frac{\sqrt{2mE}}{\hbar}l = n\pi; n = 0, \pm 1, \pm 2, \dots$$

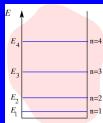


I.B. Eigenfunctions and eigenvalues: quantization

$$\Psi_{\text{II}}(x) = A \sin\left(\frac{n\pi x}{l}\right); \quad n = 1, 2, \dots \begin{cases} n = 0 & \text{no normalizable} \\ n < 0 & \text{phase factor} \end{cases}$$



- Quantization $E_n = \frac{\hbar^2}{8ml^2} n^2$
 $n = 1, 2, \dots$
- Ground state $\Rightarrow E_1$
- Spacing $\Rightarrow E_n = n^2 E_1$



I.B. Eigenfunctions and eigenvalues: normalization

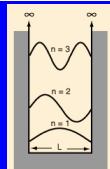
■ Normalization

$$\int_{-\infty}^{+\infty} |\psi(x)|^2 dx = 1$$

$$\underbrace{\int_{-\infty}^0 |\psi_I(x)|^2 dx}_{0} + \underbrace{\int_0^l |\psi_{II}(x)|^2 dx}_{0} + \underbrace{\int_l^{+\infty} |\psi_{III}(x)|^2 dx}_{0} = 1$$

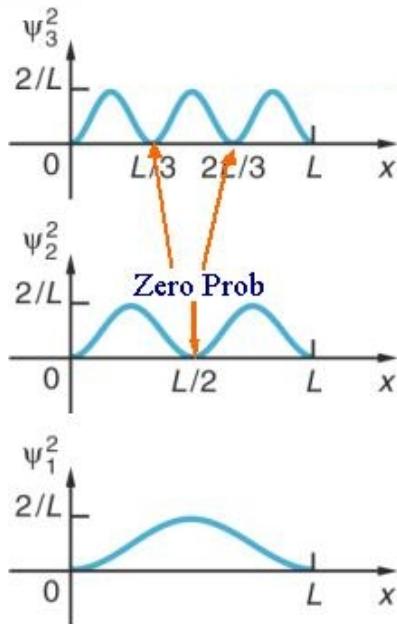
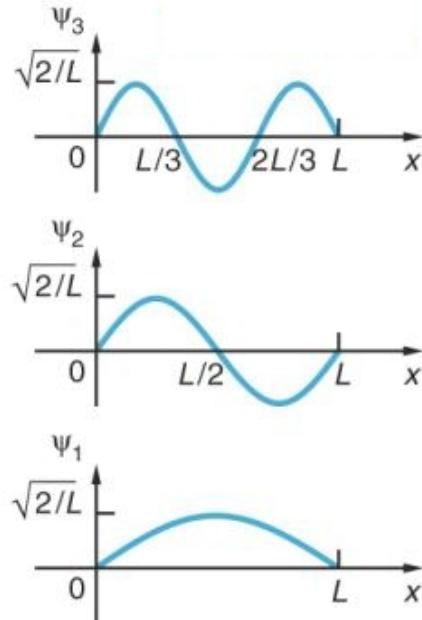
$$|A|^2 \int_0^l \sin^2\left(\frac{n\pi x}{l}\right) dx = 1 \Rightarrow |A| = \sqrt{\frac{2}{l}}$$

$$\psi_{II}(x) = \sqrt{\frac{2}{l}} \sin\left(\frac{n\pi x}{l}\right)$$



I.B. Eigenfunctions and eigenvalues: gráficas

I. Particle in a box



- Nodes

$$\Psi(x_i) = 0$$



I.C. Classical limit

Ex. electron vs melon

■ Electron in a box

$$m = 9.1 \cdot 10^{-31} \text{ kg}$$

$$l = 10 \text{ Å}$$

$$E_n = 6 \cdot 10^{-20} n^2 \text{ J}$$

$$\Delta E \sim 10^{-20} \text{ J}$$

■ Melon in a box

$$m = 1 \text{ kg}$$

$$l = 1 \text{ m}$$

$$E_n = 4 \cdot 10^{-68} n^2 \text{ J}$$

$$\Delta E \sim 10^{-68} \text{ J}$$

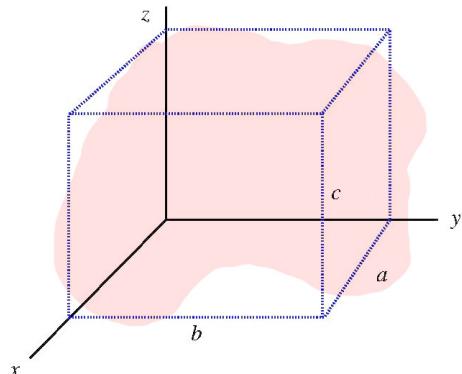
$$v = 1 \text{ m/s} \Rightarrow E = 0.5 \text{ J}$$

$$n = \sqrt{\frac{8ml^2E}{h^2}} \sim 10^{33}$$



3D box

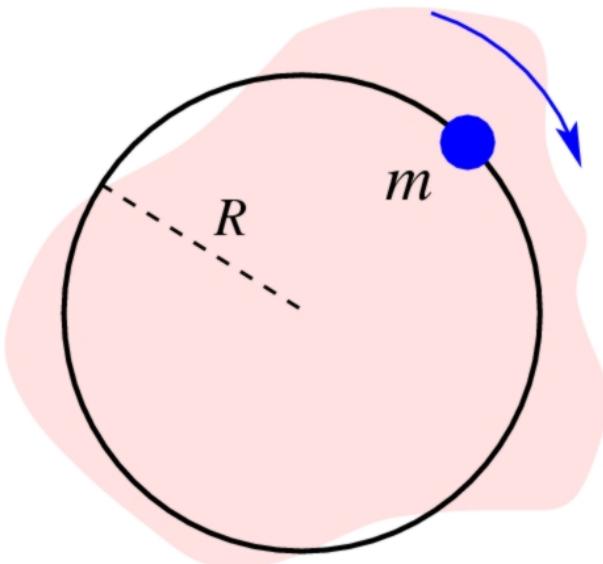
$$\hat{H}_{3D}(x, y, z) = -\frac{\hbar^2}{2m} \left(\frac{d^2}{dx^2} + \frac{d^2}{dy^2} + \frac{d^2}{dz^2} \right) \begin{cases} 0 \leq x \leq a \\ 0 \leq y \leq b \\ 0 \leq z \leq c \end{cases}$$



- $\hat{H}_{3D}\Psi_{3D} = E_{3D}\Psi_{3D}$
- $\hat{H}_{3D}(x, y, z) = \hat{H}_{1D}(x) + \hat{H}_{1D}(y) + \hat{H}_{1D}(z)$
- $\Psi_{3D}(x, y, z) = \Psi_{1D}(x)\Psi_{1D}(y)\Psi_{1D}(z)$
- $\hat{H}_{1D}(\alpha)\Psi_{1D}(\alpha) = E_{1D}\Psi_{1D}(\alpha)$ ($\alpha = x, y, z$)
- $E_{3D} = \frac{n_x^2 h^2}{8ma^2} + \frac{n_y^2 h^2}{8mb^2} + \frac{n_z^2 h^2}{8mc^2}$ $n_x, n_y, n_z = 1, 2, \dots$



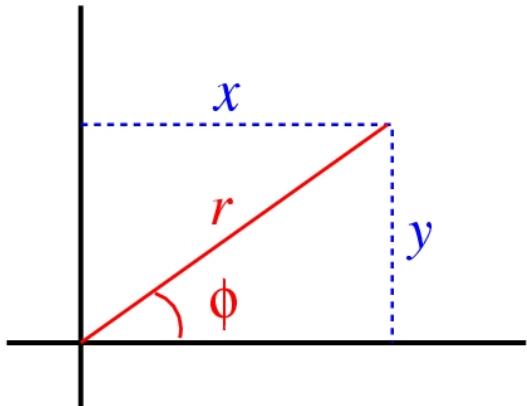
II.A. Model



- Cartesian coord. (x, y)

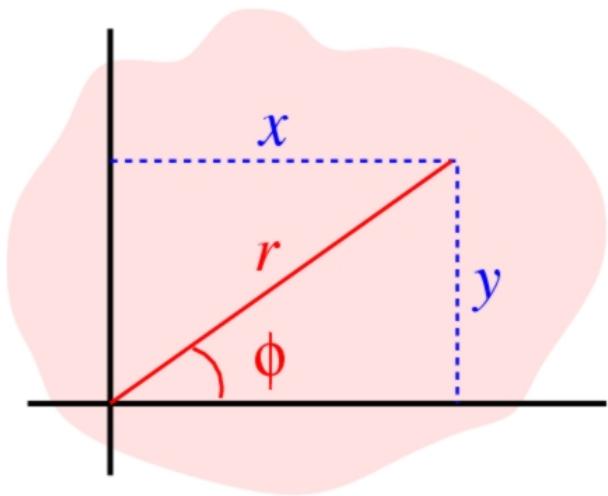
$$x^2 + y^2 = R^2$$

- Polar coord. (r, ϕ)





II.A. Model: polar coordinates



- Transformations

$$r = \sqrt{x^2 + y^2} \quad \phi = \arctan\left(\frac{y}{x}\right)$$

$$y = r \sin \phi \quad x = r \cos \phi$$

- Intervals

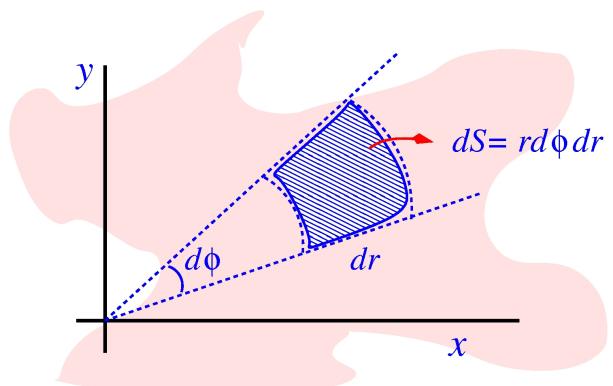
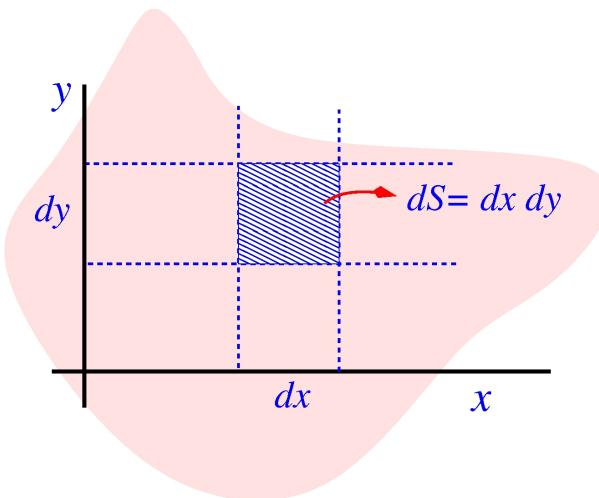
$$r \in [0, +\infty)$$

$$\phi \in [0, 2\pi)$$



II.A. Model: polar coordinates

- Differential element of surface $\Rightarrow dS = r dr d\phi$



$$\int_{-\infty}^{+\infty} \int_{-\infty}^{+\infty} f(x,y) dx dy = \int_0^{+\infty} \int_0^{2\pi} f(r,\phi) r dr d\phi$$



II.B. Eigenfunctions and eigenvalues

- Hamiltonian in polar coordinates

$$\hat{H}(x, y) = -\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} \right)$$

$$\hat{H}(r, \phi) = -\frac{\hbar^2}{2m} \left(\frac{\partial^2}{\partial r^2} + \frac{1}{r} \frac{\partial}{\partial r} + \frac{1}{r^2} \frac{\partial^2}{\partial \phi^2} \right)$$

$$\hat{H}(\phi) = -\underbrace{\frac{\hbar^2}{2mR^2}}_I \frac{\partial^2}{\partial \phi^2}$$

- Schrödinger's equation

$$-\frac{\hbar^2}{2I} \frac{\partial^2 \Psi(\phi)}{\partial \phi^2} = E \Psi(\phi)$$



II.B. Eigenfunctions and eigenvalues

- Independent solutions

$$\psi_1(\phi) = e^{k\phi}$$

$$\frac{\partial^2 \psi_1}{\partial \phi^2} = k^2 \psi_1$$

$$\psi_2(\phi) = e^{-k\phi}$$

$$\frac{\partial^2 \psi_2}{\partial \phi^2} = k^2 \psi_2$$

$$k = \sqrt{\frac{2IE}{\hbar^2}} i$$

$$\psi(\phi) = A e^{im\phi} + B e^{-im\phi} \quad m = \sqrt{\frac{2IE}{\hbar^2}}$$

II.B. Eigenfunctions and eigenvalues: degeneracy

- Single valued function $\Rightarrow \Psi(\phi) = \Psi(\phi + 2\pi)$

$$A e^{im\phi} + B e^{-im\phi} = A e^{im(\phi+2\pi)} + B e^{-im(\phi+2\pi)}$$

$$A e^{im\phi} (1 - e^{2\pi m i}) + B e^{-m\phi i} (1 - e^{-2\pi m i}) = 0$$

$$\Rightarrow 1 - e^{-2\pi m i} = e^{-2\pi m i} (e^{2\pi m i} - 1) = -e^{-2\pi m i} (1 - e^{2\pi m i})$$

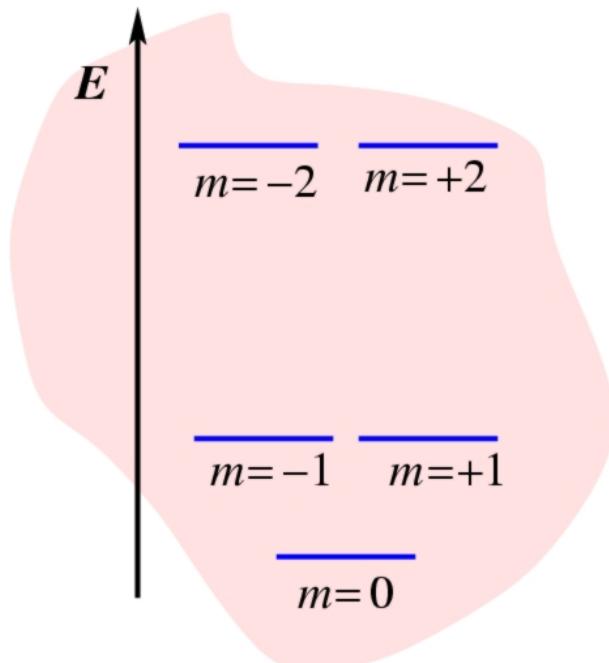
$$A e^{im\phi} (1 - e^{2\pi m i}) - B e^{-im\phi} e^{-2\pi m i} (1 - e^{2\pi m i}) = 0$$

$$\underbrace{(1 - e^{2\pi m i})}_{0} (A e^{im\phi} - B e^{-im\phi} e^{-2\pi m i}) = 0$$

$$\cos(2m\pi) + i \sin(2m\pi) = 1 \Rightarrow m = 0, \pm 1, \pm 2, \dots$$

II.B. Eigenfunctions and eigenvalues: degeneracy

- Energy quantization $\Rightarrow E_m = \frac{\hbar^2 m^2}{2I}$



- Degeneracy

$$E_{-1} = E_1$$

$$E_{-2} = E_2$$

.....



II.C. Angular momentum

- Angular momentum operator \Rightarrow If particle moves in the xy plane then $L = l_z$

$$l_z = xp_y - yp_x \rightarrow \hat{l}_z = \frac{\hbar}{i} \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right) = \rightarrow \hat{l}_z = \frac{\hbar}{i} \frac{\partial}{\partial \phi}$$

- Eigenfunctions and eigenvalues

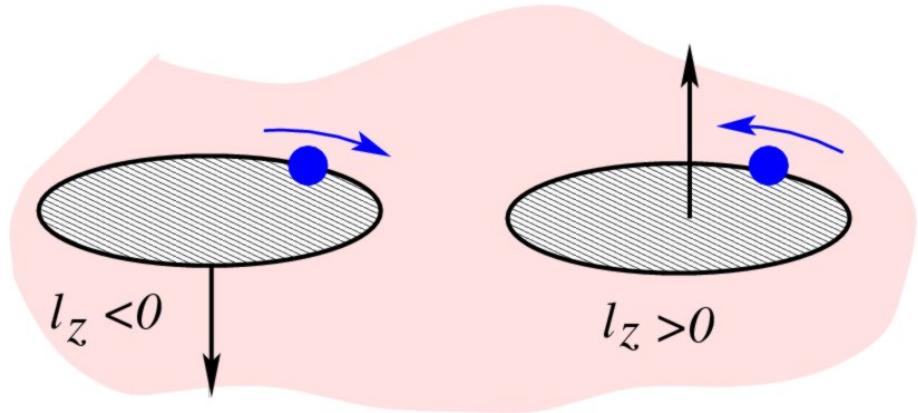
$$\begin{aligned}\hat{l}_z \psi(\phi) &= \frac{\hbar}{i} \frac{\partial}{\partial \phi} \left(A e^{im\phi} + B e^{-im\phi} \right) = m\hbar \left(A e^{im\phi} - B e^{-im\phi} \right) \\ &= l_z \left(A e^{im\phi} + B e^{-im\phi} \right)\end{aligned}$$

$$B = 0 \Rightarrow l_z = m\hbar$$

$$\psi(\phi) = A e^{im\phi} \quad m = 0, \pm 1, \pm 2 \dots$$



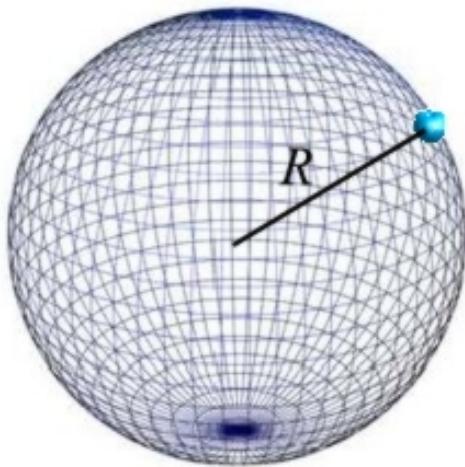
II.C. Angular momentum



- Normalization $\Rightarrow \int_0^{2\pi} |\psi(\phi)|^2 d\phi = 1 \Rightarrow |A| = \frac{1}{\sqrt{2\pi}}$
- Probability density $\Rightarrow \psi(\phi) = \frac{1}{\sqrt{2\pi}} e^{im\phi} \Rightarrow |\psi(\phi)|^2 = \frac{1}{2\pi}$



III.A. Model



- Free motion on the surface

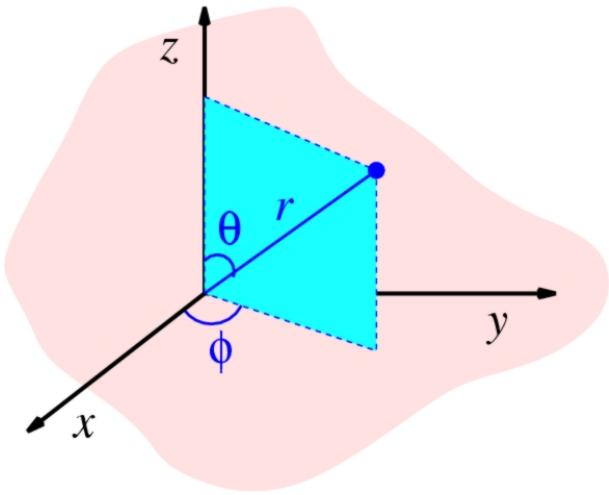
$$V(x, y, z) = 0$$

- Simple atomic model
- Cartesian coordinates

$$x^2 + y^2 + z^2 = R^2$$



III.A. Model: Spherical polar coordinates



- Spherical polar coordinates

$$x = r \sin \theta \cos \phi$$

$$y = r \sin \theta \sin \phi$$

$$z = r \cos \theta$$

- Intervals $r \in [0, \infty)$, $\phi \in [0, 2\pi)$, $\theta \in [0, \pi)$

- Differential element of volume $\Rightarrow d\tau = r^2 \sin \theta dr d\theta d\phi$

$$\iiint_{-\infty}^{+\infty} f(x, y, z) dx dy dz = \int_0^{+\infty} \int_0^\pi \int_0^{2\pi} f(r, \theta, \phi) r^2 \sin \theta dr d\theta d\phi$$



III.B. Energy and angular momentum

- Angular momentum

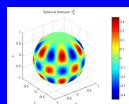
$$\hat{L}^2(\theta, \phi) = -\hbar^2 \left(\frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} + \frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \sin \theta \frac{\partial}{\partial \theta} \right)$$

$$\hat{L}^2 Y_l^m(\theta, \phi) = L^2 Y_l^m(\theta, \phi) \rightarrow L^2 = l(l+1)\hbar^2 \rightarrow l = 0, 1, 2, \dots$$

- Energy $\Rightarrow \hat{H} = \underbrace{\frac{\hat{L}^2}{2mR^2}}_I \Rightarrow E = l(l+1)\frac{\hbar^2}{2I}$
- z component angular momentum

$$\hat{L}_z = \frac{\hbar}{i} \frac{\partial}{\partial \phi} \rightarrow \hat{L}_z Y_l^m(\theta, \phi) = L_z Y_l^m(\theta, \phi)$$

$$L_z = m\hbar \quad m = -l, -l+1, \dots, 0, \dots, l-1, l$$



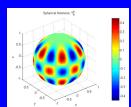
III.C. Spherical Harmonics

- Spherical Harmonics $\Rightarrow Y_l^m(\theta, \phi) = \Theta_{l,m}(\theta)\Phi_m(\phi)$

$$\Phi_m(\phi) = \frac{1}{\sqrt{2\pi}} e^{im\phi}$$

$\Theta_{l,m}(\theta) \Rightarrow$ Associated Legendre's functions

l	m	$Y_l^m(\theta, \phi)$
0	0	$1/\sqrt{4\pi}$
1	0	$\sqrt{3/4\pi} \cos \theta$
± 1	$\pm \sqrt{3/8\pi} \sin \theta e^{\pm i\phi}$	



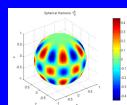
III.C. Spherical Harmonics: Probability density

 $l=0, (m=0)$

 $l=1, (m=0, |m|=1)$

 $l=2, (m=0, |m|=1, |m|=2)$

 $l=3, (m=0, |m|=1, |m|=2, |m|=3)$

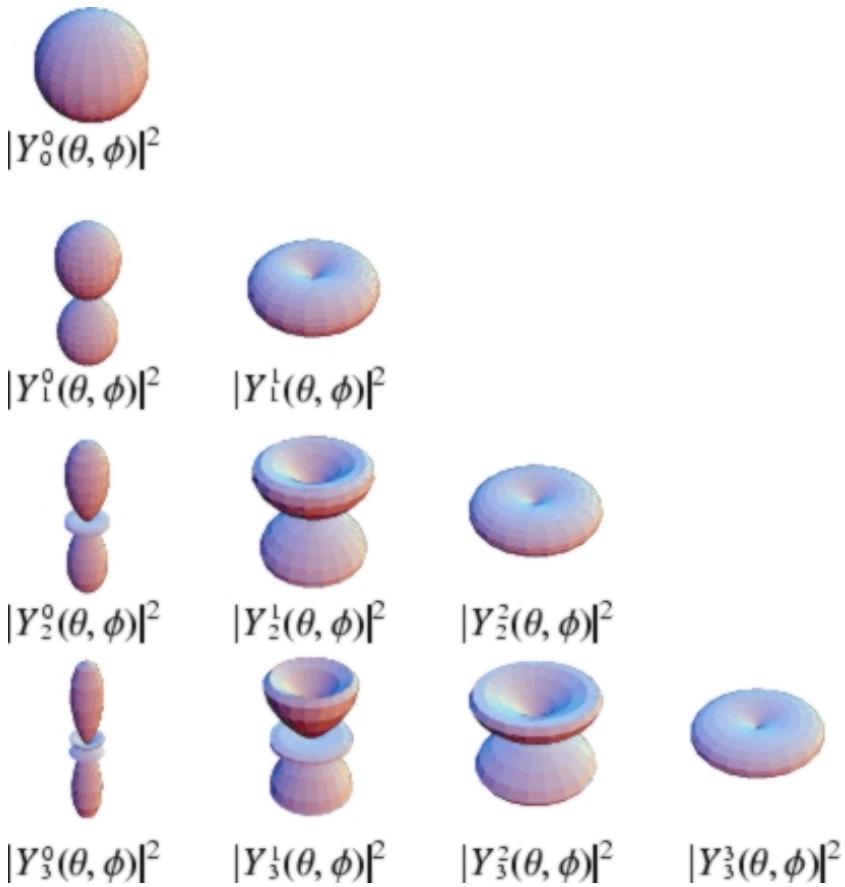
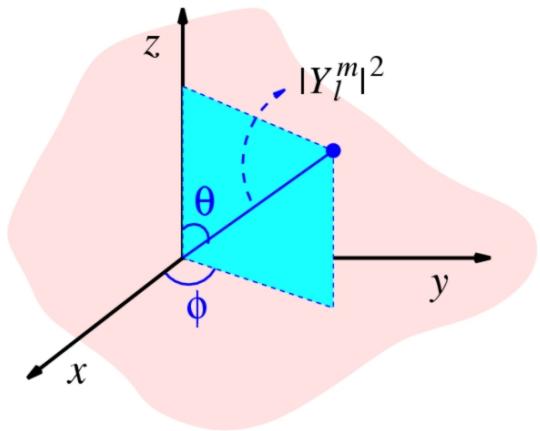


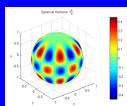
III.C. Spherical Harmonics: polar representations

USEFUL QUANTUM MODELS IN CHEMISTRY

III. Particle on a sphere

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III.C. Spherical Harmonics: notation

l	Symbol	
0	s	sharp
1	p	principal
2	d	diffuse
3	f	fundamental
4	g	alphabetical order
:	:	:
$m \Rightarrow$ subscript if $l \neq 0$		

■ Examples

$$Y_0^0 \rightarrow s$$

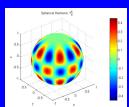
$$Y_1^1 \rightarrow p_1$$

$$Y_2^0 \rightarrow d_0$$

$$Y_1^0 \rightarrow p_0$$

$$Y_1^{-1} \rightarrow p_{-1}$$

$$Y_2^2 \rightarrow d_2$$



III.C. Spherical Harmonics: ¿ \hat{L}_x, \hat{L}_y ?

USEFUL QUANTUM MODELS IN CHEMISTRY

III. Particle on a sphere

- $[\hat{L}_x, \hat{L}_z] \neq 0, [\hat{L}_y, \hat{L}_z] \neq 0$

