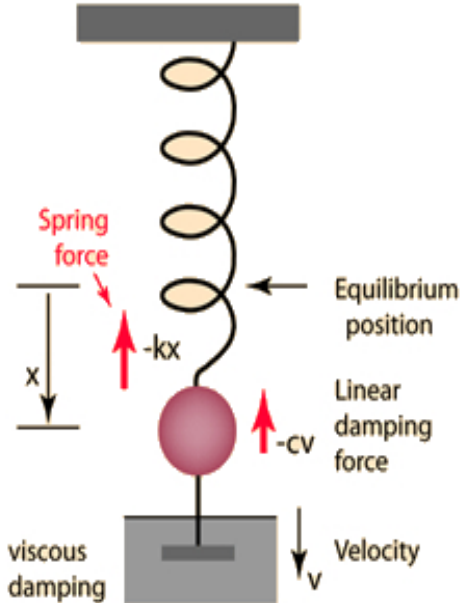




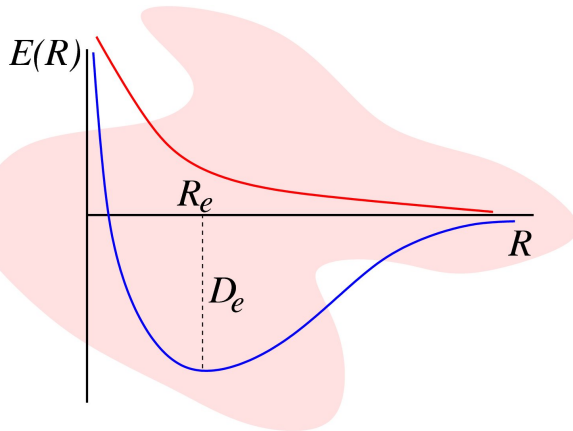
# Harmonic oscillator



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# I.A. Parabolic approximation



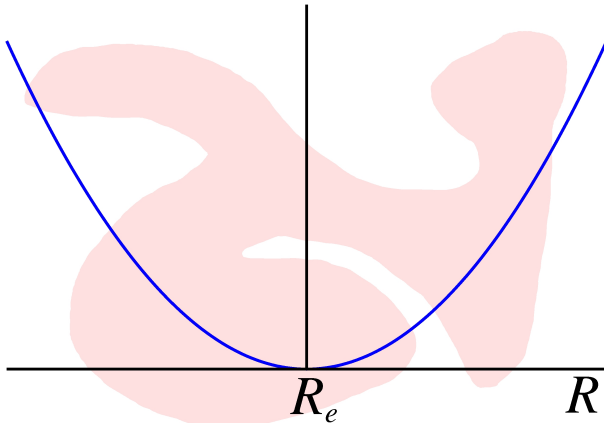
- **Unbound** states.
- **Bound** states.
  - Dissociation energy  $\Rightarrow D_e$
  - Equilibrium internuclear distance  $\Rightarrow R_e$  (bond distance)

## ■ Potential energy curve

$$V(R) = \underbrace{V(R_e)}_{\text{energy origin}} + \underbrace{\left(\frac{\partial V}{\partial R}\right)_{R_e}}_{\text{minimum}} (R - R_e) + \frac{1}{2} \left(\frac{\partial^2 V}{\partial R^2}\right)_{R_e} (R - R_e)^2 + \frac{1}{6} \left(\frac{\partial^3 V}{\partial R^3}\right)_{R_e} (R - R_e)^3 + \dots$$



# I.A. Parabolic approximation



- Harmonic potential energy

$$V_{\text{har}}(R) = \frac{1}{2} \underbrace{\left( \frac{\partial^2 V}{\partial R^2} \right)_{R_e}}_{\substack{k \rightarrow \text{force} \\ \text{constant}}} \underbrace{(R - R_e)^2}_q$$

- $\lim_{R \rightarrow \infty} V_h(R) = \infty \neq D_e$
- Minimum at  $R = R_e$
- $\lim_{R \rightarrow 0} V_h(R) = \frac{1}{2} k R_e^2 < \infty$



Hooke's law

$$F = - \frac{dV_{\text{har}}}{dq} = -kq$$

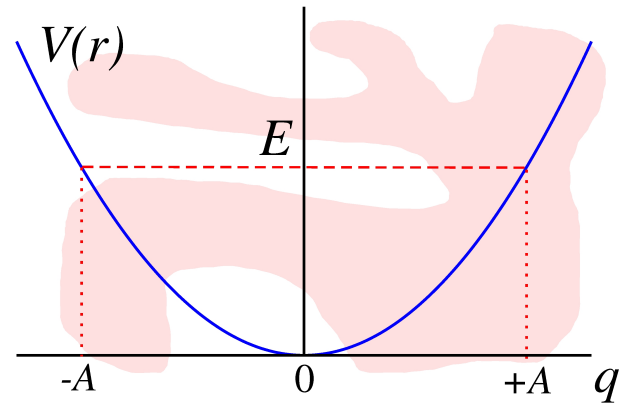


## II.A. Position

- Classical energy

$$E = \frac{p^2}{2\mu} + \frac{1}{2}kq^2 = \frac{1}{2}kA^2$$

$$\mu = \frac{m_A m_B}{m_A + m_B}$$



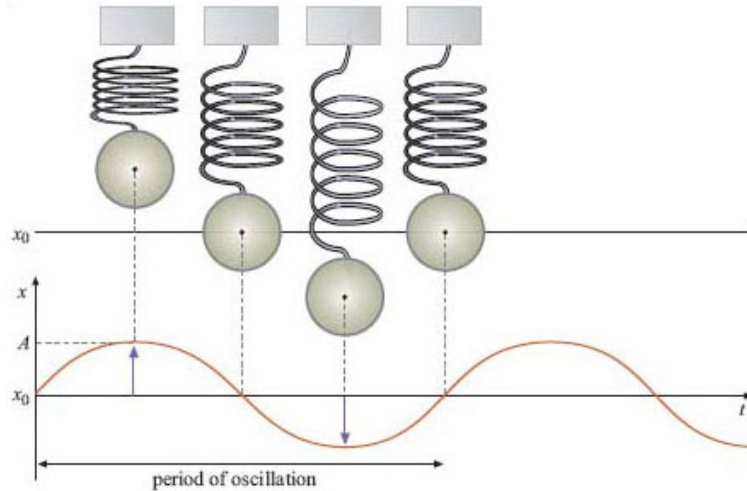
$$F = -kq = \mu a \rightarrow -kq = \mu \frac{d^2 q}{dt^2}$$

$$\downarrow q(0)=0$$

$$q(t) = A \sin(\omega t) \rightarrow \omega = 2\pi\nu = \sqrt{k/\mu}$$



## II.A. Position

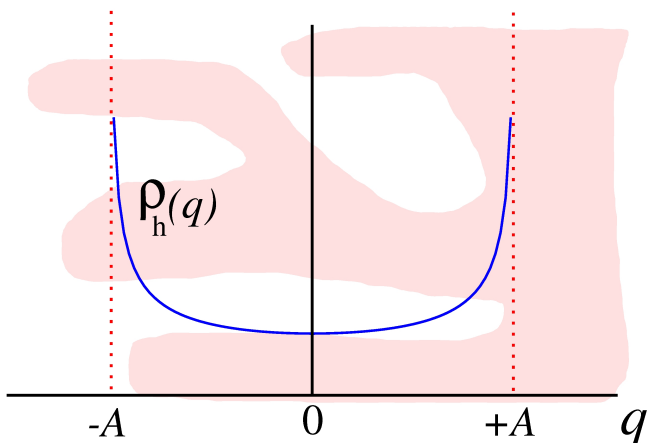


- $\nu = \frac{1}{2\pi} \sqrt{k/\mu} \Rightarrow$  Frequency ( $\text{s}^{-1}$ )
- $\omega = 2\pi\nu \Rightarrow$  Angular frequency ( $\text{s}^{-1}$ )
- $T = 1/\nu \Rightarrow$  Period (s)



## II.B. Linear momentum

$$p(t) = \mu \dot{q}(t) = \mu A \omega \cos(\omega t) = \mu \omega \sqrt{A^2 - q^2(t)}$$



$$\rho(q) \propto \frac{1}{p(t)}$$

↓

$$\rho(q) = \frac{1}{\pi \sqrt{A^2 - q^2(t)}}$$

$$\rho(p) = \frac{1}{\pi \sqrt{p_{\text{máx}}^2 - p^2(t)}}$$



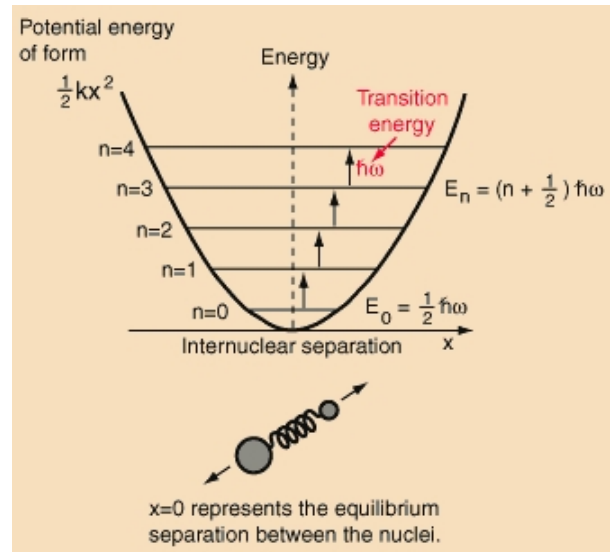
## III.A. Energy

### ■ Hamiltonian

$$\hat{H}(q) = -\frac{\hbar^2}{2\mu} \frac{d^2}{dq^2} + \frac{1}{2}kq^2 \rightarrow \hat{H}(q)\phi_n(q) = E_n\phi_n(q)$$

⇒ Eigenvalues

- $E_n = (n + \frac{1}{2}) \underbrace{h\nu}_{\hbar\omega}$ ,  $n = 0, 1, 2, \dots$
- Minimal energy  $\rightarrow E_0 = \frac{1}{2}h\nu \rightarrow$   
Zero point energy
- $\Delta E = E_{n+1} - E_n = h\nu \rightarrow$   
vibrational quantum



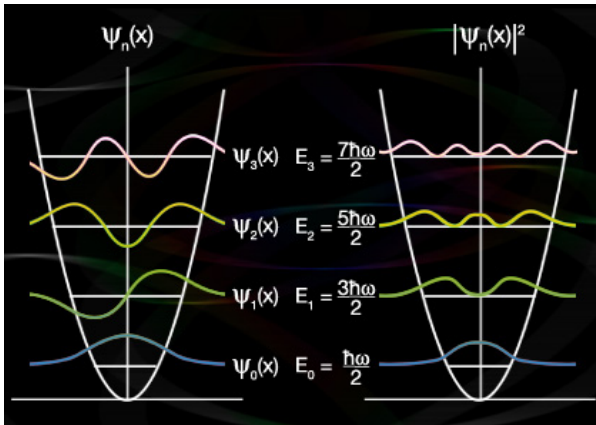


# III.B. Eigenfunctions



- Eigenfunctions  $\Rightarrow$ 

$$\begin{cases} \phi_n(q) = (2^n n!)^{-1/2} \left(\frac{\alpha}{\pi}\right)^{1/4} e^{-\alpha x^2/2} H_n(\sqrt{\alpha}x) \\ \text{Hermite's polynomials} \Rightarrow H_n(z) = (-1)^n e^{z^2} \frac{d^n e^{-z^2}}{dz^n} \\ \text{Parity} \Rightarrow n \text{ even, } n \text{ odd} \end{cases}$$



$$\phi_0(q) = \left(\frac{\alpha}{\pi}\right)^{1/4} e^{-\alpha q^2/2}$$

$$\phi_1(q) = \left(\frac{4\alpha^3}{\pi}\right)^{1/4} q e^{-\alpha q^2/2}$$

$$\phi_2(q) = \left(\frac{\alpha}{4\pi}\right)^{1/4} (1 - 2\alpha q^2) e^{-\alpha q^2/2}$$

$$\alpha = \frac{2\pi\nu\mu}{\hbar}$$

○ Nodes  $\Rightarrow n$





## III.C. Parity

- Parity

$n = 0, 2, 4, \dots \Rightarrow \phi_n(q) = \phi_n(-q) \rightarrow$  even functions

$n = 1, 3, 5, \dots \Rightarrow \phi_n(q) = -\phi_n(-q) \rightarrow$  odd functions

---

### Parity

---

even  $\cdot$  even = even

even  $\cdot$  odd = odd

odd  $\cdot$  even = odd

odd  $\cdot$  odd = even

---

▷  $\int_{-\infty}^{+\infty} f_{\text{odd}}(q) dq = 0$

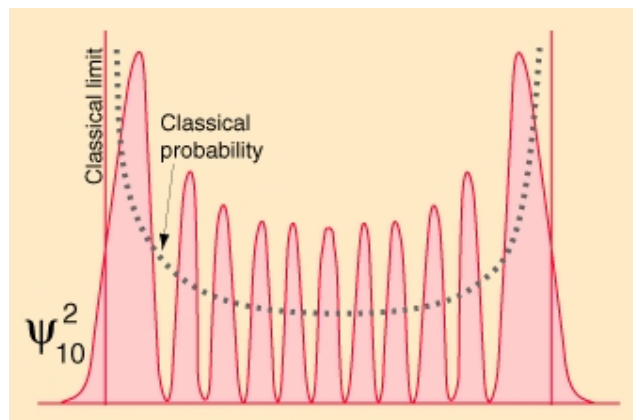
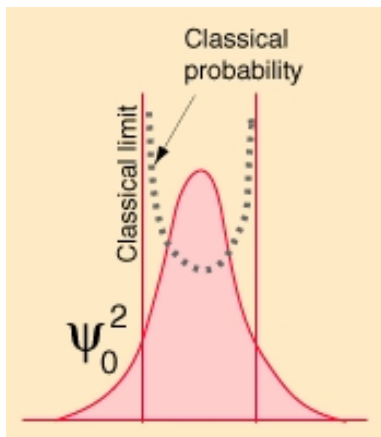
▷  $\int_{-\infty}^{+\infty} f_{\text{even}}(q) dq = 2 \int_0^{+\infty} f_{\text{even}}(q) dq$



### III.D. Classical vs quantum distributions

- Classical allowed and forbidden regions

$$E = T + V \xrightarrow{T \geq 0} E \geq V \rightarrow -A \leq q_{cl} \leq +A$$





## III.D. Classical vs quantum distributions

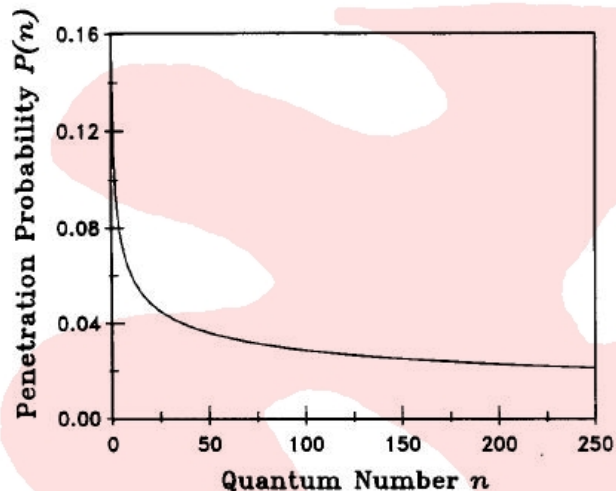


Fig. 1. The graph of the penetration probability  $P(n)$  vs quantum number  $n$  for values of  $n$  ranging from  $n=0$  to  $n=250$ .

▷ J.J. Diamond, Am. J. Phys. **60**, 912 (1992)

### ■ Problems

- No dissociation limit ( $\uparrow n$ )
- Anharmonicity  $\Rightarrow k_3 q^3 + k_4 q^4 + \dots \rightarrow \Delta E \neq \text{constant}$